

[Product Name]

Critical Care Profile (13+4)

Packing Specification

Type B: 1 Test / Disc, 10 Discs / Box.

Type B with diluent container.

Testing Instrument

Celercare V or Pointcare V chemistry analyzer

Intended Use

The Critical Care Profile (13+4) used with the Celercare V or the Pointcare V chemistry analyzer, is intended to be used for the in vitro quantitative determination of alanine aminotransferase (ALT), creatinine (CRE), urea nitrogen (BUN), glucose (GLU), potassium (K⁺), sodium (Na⁺), chloride (Cl⁻), total carbon dioxide (tCO₂), calcium (Ca), magnesium (Mg), phosphorus (P), lactate (LAC) and pH in heparinized plasma, or serum in a clinical laboratory setting or point-of-care location.

The Critical Care Profile (13+4) measurements are used in the diagnosis of water and salt metabolism disorder, acidosis, glucose metabolism and urinary system diseases.

(Principles of Testing)

The Critical Care Profile (13+4) is used to quantitatively test the concentration of the 13 biochemical indicators in the sample, which is based on the spectrophotometry. The principles are as follows:

1. Lactate (LAC)

LAC is oxidized by lactate oxidase (LOD) to pyruvate and hydrogen peroxide (H₂O₂). Peroxidase (POD) catalyzes the reaction H2O2, 4-aminoantipyrine (4-AAP) and 3, 5-dichloro-2-hydroxybenzenesulfonic acid (DHBSA) to form a red quinone imine dye.

Lactate+O₂
$$\longrightarrow$$
 Pyruvate +H₂O₂

$$2H_2O_2+4-AAP+DHBSA \xrightarrow{POD} Red Quinone imine Dye+H2O$$

The rate of formation of the red dye is proportional to the LAC concentration in the sample. The reaction is measured bichromatically at 505nm and 600nm.

2. Creatinine (CRE)

In the coupled enzyme reactions, creatinineamidohydrolase (CAH) hydrolyzes creatinine to creatine. A second enzyme, creatineamidinohydrolase (CRH), catalyzes the formation of sarcosine from creatine. Sarcosine oxidase (SAO) causes the oxidation of sarcosine to glycine, formaldehyde and hydrogen peroxide (H₂O₂). In a Trinder finish, peroxidase (POD) catalyzes the reaction among the hydrogen peroxide, 2, 4, 6-tribromo-3-hydroxybenzoic acid (TBHBA) and 4-aminoantipyrine (4-AAP) into a red quinoneimine dye. Potassium ferrocyanide and ascorbate oxidase are added to the reaction mixture to minimize the potential interference of bilirubin and ascorbic acid respectively.

Creatinine +
$$H_2O \xrightarrow{CAH}$$
 Creatine

Creatine + $H_2O \xrightarrow{CRH}$ Sarcosine + Urea



Sarcosine +
$$H_2O + O_2 \longrightarrow SAO \longrightarrow Glycine + Formaldehyde + H_2O_2
 $H_2O_2 + TBHBA + 4-AAP \longrightarrow Red Quinoneimine Dye + $H_2O_2$$$$

Two cuvettes are used to determine the concentration of creatinine in the sample. Endogenous creatine is measured in the blank cuvette, which is subtracted from the combined endogenous creatine and the creatine formed from the enzyme reactions in the test cuvette. Once the endogenous creatine is eliminated from the calculations, the concentration of creatinine is proportional to the intensity of the red color produced. The endpoint reaction is measured as the difference in absorbance at 546 nm and 700 nm.

3. Urea Nitrogen(BUN)

In the coupled-enzyme reaction, urease hydrolyzes urea into ammonia and carbon dioxide. Upon combining ammonia withα-oxoglutarate and reduced nicotinamide adenine dinucleotide (NADH), the enzyme glutamate dehydrogenase (GLDH) oxidizes NADH to NAD⁺.

$$Urea + 2H_2O \xrightarrow{Urease} 2NH_4^+ + CO_3^{2-}$$

$$NH_4^+ + \alpha - Oxoglutarate + NADH \xrightarrow{GLDH} L-Glutamate + H_2O + NAD^+$$

The rate of change of the absorbance difference between 340 nm and 405 nm is caused by the conversion of NADH to NAD⁺ and is directly proportional to the amount of urea present in the sample.

4. Glucose (GLU)

The reaction of glucose with adenosine triphosphate (ATP) catalyzed by hexokinase (HK), produces glucose-6-phosphate (G-6-P) and adenosine diphosphate (ADP). Glucose-6-phosphate dehydrogenase (G-6-PDH) catalyzes the reaction of G-6-P into 6-phosphogluconate and the reduction of nicotinamide adenine dinucleotide phosphate (NADP+) to NADPH.

$$Glucose + ATP \xrightarrow{HK} Glucose-6-Phosphate + ADP$$

$$G-6-P + NADP^{+} \xrightarrow{G-6-PDH} 6-Phosphogluconate + NADPH+H^{+}$$

The absorbance is measured bichromatically at 340 nm and 405 nm. The production of NADPH is directly proportional to the amount of glucose present in the sample.

5. Potassium (K+)

In the coupled enzyme reaction, pyruvate kinase (PK) dephosphorylates phosphoenolpyruvate (PEP) to form pyruvate. Lactate dehydrogenase (LDH) catalyzes conversion of pyruvate to lactate. Concomitantly, NADH is oxidized to NAD⁺. The rate of change in absorbance due to the conversion of NADH to NAD⁺ is directly proportional to the amount of potassium in the sample.

Interferences from other ions are minimized with the addition of some special ingredients.

$$ADP + PEP \xrightarrow{K^+, PK} Pyruvate + ATP$$

$$Pyruvate + NADH + H^+ \xrightarrow{LDH} Lactate + NAD^+$$

6. Sodium (Na⁺)

In the enzymatic reaction, β -D-galactosidase is activated by the sodium in the sample. The activated enzyme catalyzes the reaction of o-nitrophenyl- β -D-galactopyranoside (ONPG) to o-nitrophenol and galactose.



ONPG
$$\xrightarrow{Na^+, \beta\text{-D-galactosidase}}$$
 o-Nitrophenol + Galactose

7. Chloride (Cl⁻)

The method is based on the determination of chloride-dependent activation of α -amylase activity. Deactivated α -amylase is reactivated by addition of the chloride ion. The reactivation of α -amylase activity is proportional to the concentration of chloride ion in the sample. The reactivated α -amylase converts the substrate, 2-chloro-4-nitrophenyl- β -1,4-galactopyranosylmaltoside (CNP-G2) to 2-chloro-4-nitrophenol (CNP) producing color and 1,4-galactopyranosylmaltoside. The reaction is measured bichromatically and the increase in absorbance is directly proportional to the reactivated α -amylase activity and the concentration of chloride ion in the sample.

$$CNP-G2 \xrightarrow{CI^-, \alpha-amylase} CNP+G2$$

8. Total Carbon Dioxide (tCO₂)

In the enzymatic method, the specimen is first made alkaline to convert all forms of carbon dioxide (CO₂) to bicarbonate (HCO₃⁻). Phosphoenolpyruvate (PEP) and HCO₃⁻ then react to form oxaloacetate and phosphate in the presence of phosphoenolpyruvate carboxylase (PEPC). Malate dehydrogenase (MDH) catalyzes the reaction of oxaloacetate and reduced nicotinamide adenine dinucleotide (NADH) to NAD⁺ and malate. The rate of change in absorbance due to the conversion of NADH to NAD⁺ is directly proportional to the amount of CO₂ in the sample.

PEP +
$$HCO_3$$
 Oxaloacetate + Phosphate

Oxaloacetate + $NADH + H^+ \xrightarrow{MDH} NAD^+ + Malate$

9. Calcium (Ca)

Calcium in the patient sample binds with arsenazo III to form a calcium-dye complex.

$$Ca^{2+}$$
 + Arsenazo III \longrightarrow Ca^{2+} -Arsenazo III Complex

It is an endpoint reaction. The amount of total calcium in the sample is proportional to the absorbance.

10. Magnesium (Mg)

The hexokinase (HK) activation method is described as:

$$Glucose + ATP \xrightarrow{HK, Mg^{2+}} G-6-P + ADP$$

$$G-6-P + NADP^{+} \xrightarrow{G-6-PDH} 6-Phosphogluconate + NADPH + H^{+}$$

The rate limiting reaction is the HK reaction. Magnesium from the sample activates HK, which in turn catalyzes the breaking down of glucose to form glucose-6-phosphate (G-6-P) and ADP. G-6-P reacts with nicotinamide adenine dinucleotide phosphate (NADP+) to form reduced nicotinamide adenine dinucleotide phosphate (NADPH) and 6-phosphogluconate in the presence of glucose-6-phosphate-dehydrogenase (G-6-PDH). This is a first-order rate reaction. The rate of production of NADPH is directly proportional to the amount of magnesium present in the sample. Absorbance is measured bichromatically at 340 nm and 405 nm.

11. Phosphorus (P)

The enzymatic method for the MNCHIP system uses maltose phosphorylase (MP) coupled through β



-phosphoglucomutase (β -PGM) and glucose-6-phosphate dehydrogenase (G6PDH). The amount of NADH formed can be measured as an endpoint at 340/405 nm.

Maltose +Pi
$$\xrightarrow{MP}$$
 Glucose-1-Phosphate (G-1-P)+ Glucose

Glucose-1-Phosphate (G-1-P) $\xrightarrow{\beta-PGM}$ Glucose-6-Phosphate (G-6-P)

Glucose-6-Phosphate (G-6-P)+NAD+ \xrightarrow{GGPDH} NADH+ 6-Phosphogluconate+H+

12. pH

The phenol red indicator was used to detect the acidity and alkalinity of blood under different pH conditions.

13. Alanine Aminotransferase (ALT)

ALT catalyzes the transfer of an amino group from L-alanine to α-ketoglutarate to form L-glutamate and pyruvate. Lactate dehydrogenase catalyzes the conversion of pyruvate to lactate. Concomitantly, NADH is oxidized to NAD+, as illustrated in the following reaction scheme.

L-Alanine +
$$\alpha$$
-Ketoglutarate \xrightarrow{ALT} L-Glutamate + Pyruvate
Pyruvate + NADH + H⁺ \xrightarrow{LDH} Lactate + NAD⁺

The rate of change of the absorbance difference between 340 nm and 405 nm is due to the conversion of NADH to NAD⁺ and is directly proportional to the amount of ALT present in the sample.

Principle of Operation

Refer to the Celercare V or the Pointcare V chemistry analyzer Operator's Manual, for the Principles and Limitations of the Procedure.

[Description of Reagents]

Each Critical Care Profile (13+4) contains lyophilized test-specific reagent beads. A lyophilized blank reagent bead includes in each disc for a judgment of error 0233.

Type B is the reagent disc with diluent container.

Calibration information is included in barcode code. Please check it on the label.

The component of each Critical Care Profile (13+4) is as follows (after redissolution):

Component	Quantity
Lactate assay reagent	13.5 μL
Creatinine assay reagent	$13.5~\mu L$
Urea assay reagent	13.5 μL
Glucose assay reagent	6.6 µL
Potassium assay reagent	13.5 μL
Sodium assay reagent	13.5 μL
Chloride assay reagent	13.5 μL



Total Carbon dioxide assay reagent	6.6 μL
Magnesium assay reagent	13.5 μL
Phosphorus assay reagent	13.5 μL
pH assay reagent	13.5µL
Alanine aminotransferase assay reagent	13.5 μL
Calcium assay reagent	9.7 μL
Stabilizer	Appropriate amount

[Storage]

Store reagent discs in their sealed pouches at a temperature of 2-8°C (36-46°F). Do not expose opened or unopened discs to direct sunlight or temperatures exceeding 32°C (90°F). Reagent discs may be used until the expiration date indicated on the package, which is also encoded in the unique code printed on the sealing pouch.

A torn or damaged pouch may allow moisture to reach the unused disc, adversely affecting its performance. Therefore, do not use any disc from a damaged pouch.

Sample Requirements

Sample collection techniques are described in the "Sample requirement" section of the Celercare V or the Pointcare V-chemistry analyzer Operator's Manual.

The required sample usage is $100 \ \mu L$ of lithium heparin plasma, serum or quality controls.

At the same time, it is necessary to carry out the test within 60 minutes. Before taking the test, shake the lithium heparin blood collection tube gently upside down several times.

The glucose concentration is affected by the patient's feeding time and the storage environment after the sample is collected. In order to accurately measure glucose, a sample of the patient should be taken after at least 12 hours of fasting. For uncentrifuged samples stored at room temperature, the glucose concentration is reduced by about $5-12~{\rm mg}\,/\,{\rm dL}$ in 1 hour.

The test was started within 10 minutes after transferring the sample to the reagent disc.

【Interfering Substances】

Studies on known drugs or chemicals have found that when the interfering substances contained in the sample exceed the contents in the table below, the final test results are affected.

			Interfering subst	ances concenti	ration (\leq)				
A 1	Bilirubin	Intralipid	Hemoglobin	Vitamin C	Pyruvate	Creatine	NH ₄ Cl	Ca^{2+}	Mg^{2+}
Analyte	mg/dL	mg/dL	mg/dL	mg/dL	mmol/L	μmol/L	mmol/L	mmol/L	mmol/L
LAC	22.5	120	500	10	1				
CRE	40	1050	500	25		600			
BUN	25	600	1000				1		
GLU	40	600	1000	50					
K^+	16	150	50	75					
Na^+	10	150	50	75					
Cl-	18	210	50	75					



.00	4.5	525	250	7.5				
tCO_2	45	525	250	75		 		
Ca	180	210	200	75		 		3
Mg	120	140	50			 	2	
P	45	525	100	27		 		
pН	10	150	50	25		 		
ALT	40	600	50	50	1	 		

[Procedure]

■ Materials Provided

Critical Care Profile (13+4)

Celercare V or Pointcare V chemistry analyzer

Please tear off the aluminum strip before using Type B.

Transfer pipettes (fixed volume $100 \mu L$ for sample) and tips

■ Test Procedure

The complete sample collection and step-by-step operating procedures are detailed in the Operator's Manual for the Celercare V or Pointcare V chemistry analyzer.

■ Calibration

Each batch of reagent is calibrated using Randox standard serum to obtain the disc-specific calibration parameters before shipment.

The calibration parameters stored in the two-dimensional code printed on the sealed pouch are provided to analyzer at the time of scanning the code.

Refer to the Operator's Manual for specific information.

■ Quality Control

Refer to Operator's Manual of the Celercare V or the Pointcare V chemistry analyzer. Performance of the Celercare V or the Pointcare V chemistry analyzer can be verified by running controls. For a list of approved quality control materials with acceptance ranges, please consult the manual.

If control results are out of range, repeat one time. If still out of range, call MNCHIP customer service or local distributers for technical support. Do not report the results if controls are outside their labeled limits.

■ Results

The Celercare V or the Pointcare V chemistry analyzer automatically calculates and prints the analyte concentrations in the sample. Details regarding endpoint and rate reaction calculations can be found in the Celercare V or the Pointcare V chemistry analyzer Operator's Manual.

■ Normal Reference Ranges

These ranges are provided as a guideline only. It is recommended that your office or institution establish normal ranges for your particular patient population.

Analyte	SI Units	Common Units
LAC	Dog: 0.5 ~ 3mmol/L;	Dog: 4.53 ~ 27mg/dL;
LAC	Cat: 0.5 ~ 3mmol /L	Cat: 4.53 ~ 27mg/dL
CRE	Dog: 27 ~ 149μmol/L;	Dog: 0.3 ~ 1.7mg/dL;
	Cat: 27 ~ 223µmol/L	Cat: 0.3 ~ 2.5mg/dL



DIIN	Dog: 2.5 ~ 11.5mmol/L	Dog: 7 ~ 32mg/dL
BUN	Cat: 3.6 ~ 15.5mmol/L	Cat: $10 \sim 43 \text{mg/dL}$
CLU	Dog: 3.89 ~ 7.95mmol/L	Dog: 70 ~ 143mg/dL;
GLU	Cat: 4.11 ~ 8.84mmol/L	Cat: 74 ~ 159mg/dL
K ⁺	Dog: 3.7 ~ 5.8mmol/L;	Dog: 3.7 ~ 5.8mmol/L;
K.	Cat: 3.7 ~ 5.8mmol/L	Cat: 3.7 ~ 5.8mmol/L
$\mathrm{Na}^{\scriptscriptstyle +}$	Dog: 138 ~ 160mmol/L;	Dog: 138 ~ 160mmol/L;
Na*	Cat: 142 ~ 164mmol/L	Cat: 142 ~ 164mmol/L
Cl ⁻	Dog: 106 ~ 130mmol/L;	Dog: 106 ~ 130mmol/L;
CI	Cat: 100 ~ 126mmol/L	Cat: 100 ~ 126mmol/L
tCO_2	Dog: 12 ~ 27mmol/L;	Dog: 12 ~ 27mmol/L;
ιCO_2	Cat: 15 ~ 24mmol/L	Cat: 15 ~ 24mmol/L
Ca	Dog: 1.98 ~ 2.95mmol/L;	Dog: 7.9 ~ 11.8mg/dL;
Ca	Cat: 1.95 ~ 2.95mmol/L	Cat: 7.8 ~ 11.8mg/dL
Ma	Dog: 0.6 ~ 1.09mmol/L;	Dog: $1.5 \sim 2.6 \text{ mg/dL}$;
Mg	Cat: 0.7 ~ 1.21mmol/L	Cat: 1.7 ~ 2.9 mg/dL
P	Dog: 0.81 ~ 2.2mmol/L;	Dog: $2.5 \sim 6.8 \text{mg/dL}$;
r	Cat: 1 ~ 2.74mmol/L	Cat: 3.1 ~ 8.5mg/dL
pН	7.25-7.55	7.25-7.55
AIT	Dog: 10 ~ 140U/L;	Dog: 10 ~ 140U/L;
ALT	Cat: 8.2 ~ 123U/L	Cat: 8.2 ~ 123U/L

【Interpretation of Results】

Physiological interferents, such as hemolysis, icterus, and lipemia, can cause changes in the reported concentrations of certain analytes. Sample indices are printed at the bottom of each printout to inform the operator about any abnormalities in the sample. The operator should take care to avoid hemolysis caused by improper blood collection techniques.

The Celercare V or the Pointcare V chemistry analyzer suppresses any results that are affected by >10% interference from hemolysis, lipemia or icterus. "HEM", "LIP", or "ICT" respectively, is printed on the printout in place of the result.

For the same sample, the potassium result of using plasma is 0.2 - 0.5 mmol/L lower than those using serum. The potassium assay is a coupled pyruvate kinase (PK) / lactate dehydrogenase (LDH) assay. Therefore, in cases of extreme muscle trauma or highly elevated levels of creatine kinase (CK), The Celercare V or the Pointcare V chemistry analyzer may report a falsely elevated potassium (K⁺) value. In such cases, unexpected high potassium recoveries need to be confirmed utilizing a different methodology.

Any result for a particular test that exceeds the assay range should be analyzed by another approved test method or sent to a referral laboratory. Do not dilute the sample and run it again on the Celercare V or the Pointcare V chemistry analyzer.

[Limitations of Procedure]

The Critical Care Profile (13+4) should be used with the Celercare V or the Pointcare V chemistry analyzer, and is just used for in vitro diagnosis (IVD).

As with any diagnostic test procedure, all other test procedures including the clinical status of the



patient, should be considered prior to final diagnosis.

[Performance Characteristics]

Accuracy

Analyte	The relative deviation or absolute deviation should meet the following requirements
LAC	B% ≤ 15.0%
CRE	$B\% \le 10.0\%$
BUN	$B\% \le 15.0\%$
GLU	$B\% \le 20.0\%$
\mathbf{K}^{+}	$B\% \le 15.0\%$
Na^+	$B\% \le 15.0\%$
Cl-	$B\% \le 15.0\%$
tCO_2	$B\% \le 10.0\%$
Ca	$B\% \le 5.0\%$
Mg	$B\% \le 15.0\%$ or Absolute deviation ≤ 0.2 mmol/L
P	$B\% \le 10.0\%$
pН	$B\% \le 10.0\%$
ALT	$B\% \le 15.0\%$

Batch precision

Analyte	Coefficient of variation (≤ *)	
LAC	5.0%	
CRE	5.0%	
BUN	5.0%	
GLU	5.0%	
\mathbf{K}^{+}	5.0%	
Na^+	5.0%	
Cl ⁻	5.0%	
tCO_2	5.0%	
Ca	3.0%	
Mg	5.0%	
P	5.0%	
рН	0.5%	
ALT	5.0%	

Inter batch precision

Analyte	Relative Range (≤ *)
LAC	10.0%
CRE	10.0%



BUN	10.0%	
GLU	10.0%	
\mathbf{K}^{+}	10.0%	
Na ⁺	10.0%	
Cl ⁻	10.0%	
tCO_2	10.0%	
Ca	5.0%	
Mg	10.0%	
P	10.0%	
рН	0.5%	
ALT	10.0%	

Dynamic Ranges

Analyte	Dynamic Ranges
LAC	0 ~ 9mmol/L
CRE	$20 \sim 2000 \mu \text{mol/L}$
BUN	$0.9 \sim 35.7$ mmol/L
GLU	1 ~ 35 mmol/L
K^+	1 ~ 8 mmol/L
Na^+	90 ~ 170mmol/L
Cl ⁻	60 ~ 140mmol/L
tCO_2	10 ~ 35mmol/L
Ca	0.5~4mmol/L
Mg	$0.2 \sim 1.6$ mmol/L
P	0.2 ~ 7mmol/L
рН	7.20~8.00
ALT	5 ~ 1500U/L

Notes

Used reagent discs contain animal body fluids. It is essential to follow good laboratory safety practices when handling and disposing of these used discs. For instructions on cleaning biohazardous spills, refer to the Celercare V or Pointcare V chemistry analyzer Operator's Manual.

The reagent discs are made of plastic and may crack or chip if dropped. Never use a disc that has been dropped, as it may spray biohazardous material throughout the interior of the analyzer.

Reagent beads may contain acids or caustic substances. Operators do not come into contact with the reagent beads when following the recommended procedures. It is important to avoid ingestion, skin contact, or inhalation of the reagent beads.

(Symbols Used in Labelling)

Symbol	Explanation
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Veterinary	Veterinary use only
	Manufacturer
UDI	Unique device identifier
EC REP	Authorized representative in the European Community
\square	Use-by date
LOT	Batch code
سا	Date of manufacture
[]i	Consult instructions for use
2°C. 8°C	Limit of temperature
(((((((((((((Do not re-use

[Manufacturer]



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Ver 1.1